

Faculty of Physics, AG-CMP, University of Vienna







general remarks (1)

- this excercise consists of 4 steps which unfold if you untar the file ammonia_flipping.tgz
 - tar zxvf ammonia_flipping.tgz
 - $\textcircled{0} \longrightarrow \mathsf{scf:} relaxed geometry of NH_3$
 - $\bigcirc \longrightarrow \mathsf{NEB}$: TS search using the Nudged Elastic Band method
 - ③ → TS_vib: get the vibrational modes of the TS (planar NH₃) to find the mode of the decay direction, to be given in the improved-dimer calculation
 - improved-dimer: TS search using the improved dimer method.
 - compare the results and runtimes using NEB and the improved dimer method

general remarks (2)

- for all calculations of this excercise, use:
 - PAW-PBE pseudopotentials (potpaw_PBE_54.tar.gz data set)
 - an orthorhombic $6 \times 7 \times 8$ unit cell
 - Γ-point only:
- KPOINTS

k-points

0

G

1 1 1

POSCAR

```
ammonia flipping
1.000000
6.000000 0.000000 0.000000
0.000000 7.000000 0.000000
0.000000 0.000000 8.000000
3 1
Selective dynamics
Direct
0.636429 0.567446 0.549205 T T T
0.500000 0.364896 0.549205 T T T
0.363571 0.567446 0.549205 T T T
0.500000 0.500000 0.500000 F F F
```

NH₃ Relaxed Geometry



- use Selective dynamics
- fix the position of N
 F F F

to avoid that the molecule drifts through the cell

INCAR

SYSTEM = Ammonia flipping IBRION = 2 NSW = 2 ALGO = N POTIM = 0.5 EDIFF = 1e-6EDIFFG = -0.01NELMIN = 5 5^{0} 10.40FF c)/c cm OUTCO

NH₃ Relaxed Geometry

- CG-algorithm for ionic relaxation
- blocked Davidson for el. relaxation
- at least 5 el. steps for each ionic step

 $E^0 = -19.4955$ eV: see OUTCAR energy without entropy= -19.50322126 energy(sigma->0) = -19.49549072

1. using the Nudged Elastic Band Method ...

G. Mills et.al. Surf. Sci. 324,305 (1995)

- generate the final state by applying σ_z (mirror plane through N at $z = \frac{1}{2}c$) to the relaxed (ground state) geometry
- generate directories for the intermediate geometries along the reaction path: 00, 01, 02,....07 (IMAGES+1)
 00: POSCAR of the initial state,
 07 (IMAGES+1): POSCAR of the final state
- Oconcatenate the POSCARs of the initial and final states: cat POSCAR_in POSCAR_fin > POSCAR_if
- generate the POSCAR files of the intermediate steps (IMAGES) along the reaction path by using the script interpolatePOSCAR interpolatePOSCAR POSCAR_if

1. ... using the Nudged Elastic Band Method

- to obtain reasonable results Selective Dynamics has to be used, again fixing the position of N (like for the scf-run)
- edit the POSCAR files in the sub-directories 00 ... 07 to apply these changes:

insert the keyword Selective above the line Direct (there must be NO space character at the beginning of this line) add T T T (F F F) at the end of the lines giving the positions of the atoms

- In vasp: the number of cores for a NEB job must be an integer multiple of the number of IMAGES
- the output of each IMAGE is written into the sub-directories 01
 ... 06, there is no output written to 00 and 07

INCAR

```
SYSTEM = Ammonia flipping

IMAGES = 6

SPRING = -5

IBRION = 2

NSW = 50

ALGO = N

POTIM = 1.0

EDIFF = 1e-6
```

NEB

- IMAGES defines the number of intermediate geometries along the reaction path
- SPRING: tangential springs to keep the images equidistant



post-processing

- after vasp has finished, collect the total energies of the IMAGES from the OUTCAR files in 01,...
- plot the calculated and interpolated energies eg. using the scripts showbarrier_gnuplot or
 - $\verb+showbarrier_xmgrace+$
- *E*[‡] is obtained from the interpolated data
- $E^{\ddagger}(NEB) = -19.2882 \text{ eV}$

Improved Dimer Method

A. Heyden et.al., J.Chem.Phys.123, 224101 (2005)

- the Improved Dimer Method is a method to optimize transition sates, by maximizing the potential energy along the unstable mode.
- it speeds up the search of the transition state significantly, as compared to the NEB method
- the number or cores can be reduced.
- to direction of the unstable mode (decay direction, = dimer axis) has to be specified POSCAR
- this direction can be found from analyzing the vibration spectrum of the TS: it corresponds to the hardest mode of the vibrations with imaginary frequecies (f/i)
- \Rightarrow as preliminary step, the vibration frequencies of the TS have to be calculated.

POSCAR

ammonia flipping 1.00000 6.00000 0.00000 0.00000 0.00000 7.00000 0.00000 0.00000 0.00000 8.00000 3 1 Direct 0.6462 0.5736 0.5000 0.5000 0.3547 0.5000 0.3538 0.5736 0.5000 0.5000 0.5000 0.5000

TS geometry

the TS of the NH_3 flipping reaction is a planar molecule:



INCAR

```
SYSTEM = Ammonia flipping

IBRION = 5

NSW = 1

ALGO = F

POTIM = 0.015

EDIFF = 1e-8

NWRITE = 3
```

vibration modes of the TS

- IBRION = 5 calculate the vibration modes, using finite differences
- POTIM: has to be chosen small enough to stay in the harmonic regime of the oscillators
- EDIFF: tight convergence criterium for the energies to obtain accurate frequencies

extract the decay mode from OUTCAR

- altogether there are 12 DOFs, leading to 12 vibrational modes.
- 3 of these modes (those with the lowest energies) are related to translational modes of the molecule through the box
- the hardest imaginary mode corresponds to the decay path from the TS:
- the last 3 columns correspond to the eigenvectors of the modes:

```
12 f/i= 21.058633 THz 132.315291 2PiTHz 702.440353 cm<sup>1</sup>
87.091535 meV
X Y Z dx dy dz
3.877200 4.015200 4.000000 0.000005 -0.000005 0.511988
3.000000 2.482900 4.000000 -0.000005 -0.000003 0.547857
2.122800 4.015200 4.000000 -0.000003 0.000002 0.511993
3.000000 3.500000 4.000000 0.000000 -0.111986
```

POSCAR

ammonia flipping 1.00000 6.00000 0.00000 0.00000 0.00000 7.00000 0.00000 0.00000 0.00000 8.00000 3 1 Direct 0.6462 0.5736 0.5000 0.5000 0.3547 0.5000 0.3538 0.5736 0.5000 0.5000 0.5000 0.5000 decay direction 0.000004 - 0.000001 0.5119900.00000 - 0.00003 0.547859-0.00004 - 0.00001 0.5119880.000000 0.000000 -0.111986

 the decay direction (dimer axis) as taken from the calculation of the vibration mode has to be added at the end of the file, separated from the atomic positions by a blank line

INCAR

SYSTEM = Ammonia flipping IBRION = 44 NSW = 100 ALGO = F EDIFF = 1e-6 EDIFFG = -0.01

- IBRION = 44 activates the improved dimer method
- use the plot-script showbarrier_gnuplot to display the collected results

• $E^{\ddagger}(i.d.) = -19.2875 \text{ eV}$



collecting the results

• (1) the E-barrier height (activation energy, ΔE) for the flipping process of NH₃ can be calculated from $\Delta E = E^{\ddagger} - E^{0}$

the results of the methods agree within 1 meV / molecule

- the 'computational cost' of both approaches can be compared by having a look at the number of cores and the CPU-time used for the calculations (both written on OUTCAR)
 - # cores: grep 'running on ' OUTCAR
 - CPU-time: grep 'Total CPU time used ' OUTCAR